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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/595,795	08/15/2006	Karl Schermanz	16785.1	6850
22913 Workman Nyde	7590 12/22/201 egger	EXAMINER		
1000 Eagle Gate Tower			DARJI, PRITESH D	
60 East South Temple Salt Lake City, UT 84111			ART UNIT	PAPER NUMBER
			1731	
			MAIL DATE	DELIVERY MODE
			12/22/2010	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)		
Office Action Occurrence	10/595,795	SCHERMANZ ET AL.		
Office Action Summary	Examiner	Art Unit		
	PRITESH DARJI	1731		
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence address		
A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING DA - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be time will apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).		
Status				
1) ☐ Responsive to communication(s) filed on <u>02 Not</u> 2a) ☐ This action is FINAL . 2b) ☐ This 3) ☐ Since this application is in condition for alloward closed in accordance with the practice under E	action is non-final. nce except for formal matters, pro			
Disposition of Claims				
4) ☐ Claim(s) 21-24 is/are pending in the application 4a) Of the above claim(s) is/are withdray 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 21-24 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/or	vn from consideration.			
Application Papers				
9) ☐ The specification is objected to by the Examine 10) ☑ The drawing(s) filed on 11 May 2006 is/are: a) ☐ Applicant may not request that any objection to the case Replacement drawing sheet(s) including the correction 11) ☐ The oath or declaration is objected to by the Examine 11.	☑ accepted or b) ☐ objected to be drawing(s) be held in abeyance. See on is required if the drawing(s) is obj	e 37 CFR 1.85(a). jected to. See 37 CFR 1.121(d).		
Priority under 35 U.S.C. § 119				
 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 				
Attachment(s) 1)	4) ☐ Interview Summary	(PTO-413)		
2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:	ate		

DETAILED ACTION

Continued Examination Under 37 CFR 1.114

A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 11/02/2010 has been entered.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 21-24 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kleemann (Investigation of... analysis) in view of Reddy (Surface... Techniques).

Regarding claims 21 and 23, Kleemann teaches a process for preparation of a catalyst composition in which monolithic cordierite honeycomb was immersed in the suspension of TiO₂ with 9% WO₃. Since there is not any other component present in the suspension, 91 wt% TiO₂ is present in the suspension. It is obvious that slurry is formed because TiO₂ and WO₃ are powders and their presence in the suspension

would form slurry. After impregnation sample is dried and it is impregnated with NH₄VO₃. Sample is dried and calcined. See 2. Experimental, 2.1. Absence of SiO₂ makes its wt% 0, therefore the limitation is met.

Kleemann does not teach that REVO4 is contacted with the TiO₂ and WO₃.

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Reddy, drawn to surface characterization of CeO_2/SiO_2 and $V_2O_5/CeO_2/SiO_2$, disclosed a method of preparing 5% $V_2O_5/CeO_2/SiO_2$ catalysts in which vanadium oxide with ammonium metavanadate (NH₄VO₃) is dissolved in aqueous oxalic acid. Then powdered support as added. The resulting material was dried and calcined. See Experimental Section, para 2 and table 1. Calcining the resultant at 973K resulted in the formation of $CeVO_4$. See pg 10967, col. 1, lines 1-12. Therefore the formation of $CeVO_4$ occurs in the process. In the process powdered support materials are added as well so they are in contact with the formed $CeVO_4$.

It would have therefore been obvious to one of ordinary skill in the art at the time of invention to have substituted CeVO₄ and SiO₂/V₂O₅/CeO₂ of Reddy for ammonium vanadate of Kleemann, motivated by the fact that the combination of vanadia (known for its redox properties) and ceria (known for its oxygen storage and release functions) gives rise to a catalyst system that may catalyze extraneous redox reactions for both selective and non-selective oxidation. See pg 10965, col. 1, lines 2-18. Kleemann uses ammonia metavanadate (NH₄VO₃), which can be replaced by cerium vanadate (CeVO₄) to yield above stated advantages.

Regarding claim 22, any difference imparted by the product by process limitations would have been obvious to one having ordinary skill in the art at the time the

invention was made because where the examiner has found a substantially similar product as in the applied prior art the burden of proof is shifted to the applicant to establish that their product is patentably distinct not the examiner to show the same process of making, see In re Brown, 173 USPQ 685, In re Fessmann, 180 USPQ 324, In re Spada, 15 USPQ2d 1655, In re Fitzgerald, 205 USPQ 594 and MPEP 2113.

Regarding claim 24, Reddy teaches method of preparing $5\% V_2O_5/CeO_2/SiO_2$ catalysts in which vanadium oxide with ammonium metavanadate (NH₄VO₃) is dissolved in aqueous oxalic acid. Then powdered support as added. The resulting material was dried and calcined. See Experimental Section, para 2, abstract and table 1.

Response to Arguments

Applicant's arguments filed on 11/02/2010 have been fully considered but they are not persuasive.

Applicant argues that from Kleemann, a skilled person would not be motivated to use rare earth metal vanadate instead of ammonium vanadate because there is absolutely no hint or indication doing so.

However, the motivation for doing so is to give rise to a catalyst system that may catalyze extraneous redox reactions for both selective and non-selective oxidation and it can be used for oxidative removal of volatile organic compounds and other *noxious emissions* (Reddy, pg 10965, col. 1, lines 2-18). Kleemann teaches catalyst for the ammonia adsorption for its use in emissions (abstract and Introduction.1). Both

catalysts are used in similar atmosphere, therefore their use in emissions was a hint to combine Kleemann and Reddy.

Applicant argues CeVO₄ indicated in Reddy is obtained only **after calcination** with increasing calcination temperature from 973 to 1073 K, and even higher temperature (pg 10967, col. 2, para 2).

However, it is unclear how calcination renders combination of Reddy with Kleemann non obvious. Instant claims are not excluding any temperature range of the process being performed in.

Applicant argues that a skilled person would not be motivated to use TiO₂ or WO₃, and would definitely not be motivated to use both TiO₂ and WO₃.

However, Kleemann (primary references) use of TiO₂ and WO₃ together (2.1. Preparation of the coated catalysts). The need of motivation is unclear when above stated limitation is taught by primary reference.

Applicant argues that Kleemann uses ammonium metavanadate to "Investigation of the ammonia adsorption on monolithic SCR catalysts,". Changing ammonia metavanadate to a rare earth vanadate would change the technology described therein to be modified in a manner that is unsatisfactory for the intended purpose of the investigation of ammonia adsorption. Therefore the proposed combination of Kleemann and Reddy is improper and against the established case law.

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However, according Kleemann active components of the catalyst are V₂O₅ and WO₃ supported on TiO₂, and surface of catalyst exhibits strongly acid properties and thus adsorbs ammonia, therefore ammonium vanadate is not key component for ammonium adsorption (pg 231, col. 2, para 2). Therefore substituting vanadates wont effect ammonia adsorption and combination is valid.

Applicant argues that an ammonium metavanadate in Kleemann to be replaced by cerium vanadate bring impermissible hindsight using the Applicant's claims as a roadmap could result in such an assertion.

However, as explained above, substituting vanadates wont effect ammonia adsorption and combination is valid. In response to applicant's argument that the examiner's conclusion of obviousness is based upon improper hindsight reasoning, it must be recognized that any judgment on obviousness is in a sense necessarily a reconstruction based upon hindsight reasoning. But so long as it takes into account only knowledge which was within the level of ordinary skill at the time the claimed invention was made, and does not include knowledge gleaned only from the applicant's disclosure, such a reconstruction is proper. See *In re McLaughlin*, 443 F.2d 1392, 170 USPQ 209 (CCPA 1971).

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to PRITESH DARJI whose telephone number is (571)270-

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5855. The examiner can normally be reached on Monday to Thursday 8:00AM EST to 6:00PM EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Jerry Lorengo can be reached on 571-272-1233. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/J.A. LORENGO/ Supervisory Patent Examiner, Art Unit 1731

/P. D./ Examiner, Art Unit 1731